CAMBRIDGE INTERNATIONAL EXAMINATIONS GCE Ordinary Level

MARK SCHEME for the October/November 2012 series

5129 COMBINED SCIENCE

5129/22 Paper 2 (Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2012 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.



			GCE O LEVEL – Octo	ber/November 2012	5129	22
1	glar bloo targ	od get				[4]
	1116	<u>.</u>				[4]
2	(a)	= 6.0	or 2.5 × 2.4 dependent)			[1] [1] [1]
	(b)	no. turns	of magnetic field	any two		
		mass of ignore:	weight size of coil, bigger magnet,	speed alone and weight	alone	[2]
3	(a)	arrow <u>ve</u>	<u>rtically</u> down (anywhere on	diagram)		[1]
	(b)	(i) Pat	beginning of path (above the	he building)		[1]
		(ii) Kat	end of path			[1]
	(c)	rate of cl change i	hange of velocity / speed n velocity / time	any 1		[1]
4	(a)	2, 8 (igno	ore correct charge)			[1]
	(b)		(divide by 10) (divide by 4) ighout			[2] [1] [1]
	(c)	ionic / el	ectrovalent			[1]
5	(a)	concrete	expands			[1]
	(b)	path / co	ncrete buckles ncrete cracks / breaks destroy path / concrete	any 1		[1]

Syllabus

Paper

	Page 3			wark Scheme	Syllabus	Paper
			G	CE O LEVEL – October/November 2012	5129	22
6	(a)	B = C =	cytopla nucleu		any 1	[3]
	(b)	con	trols / a	allows movement of chemicals out of the cell ation must match the difference.	, "", "	[1]
		wark ui	nerend	ce and explanation together]		
		difference explanate		no nucleus (in red blood cell) (cell can contain) <u>more</u> haemoglobin (cell can carry) <u>more</u> oxygen		
		difference explanate		biconcave (disc) shape large surface area (per volume) increased uptake of <u>oxygen</u> (in lung capillar increased release of <u>oxygen</u> (in tissue capill faster diffusion ignore: easier to carry oxygen		
		difference explanate		flexible / small size of red blood cell cell can pass through capillaries rapidly		[6]
7	(a)	(i) gam	nma / ŋ	,		[1]
		(ii) alph	na/α			[1]
	(b)			es into proton / p increase by 1 and n decreas : electrons change	se by 1	[1]
	(c)	use tong keep dis	stance			
		store in	lead co	ontainer / safely when not in use uit, goggles, safety gear, protective gear		[2]
8	(a)			rials are attracted to magnets / can be magnet materials are not / cannot be magnetised	etised any	1 [1]

Syllabus

Paper

	Page 4		4 Mark Scheme Syllabus	Paper
	V		GCE O LEVEL – October/November 2012 5129	22
	(b)	(i)	steel is a hard magnetic material steel retains magnetism / permanent steel hard to magnetise iron is soft magnetic material iron easily loses magnetism / temporary iron easy to magnetise	[1]
		(ii)	no difference/ none / no effect	[1]
		(iii)	more turns more current (voltage) / add more batteries	[2]
			more current (voltage) / add more batteries	[2]
	(c)	8.0	3	[1]
9	(a)	A =	= sulfuric (acid) / H ₂ SO ₄	
	()	B =	= water / H ₂ O	[9]
		C –	= copper / Cu	[3]
	(b)	eva	aporate (some of the water) / heat / boil	
	()	filte coo igno	er the crystals ol / crystallise any 1 apore initial filtration aporate to dryness max 1 mark	[2]
		eva		
	(c)	con con mal duc shir	nducts heat nducts electricity alleable ctile iny th density	
		_	norous	[2]
10			a single seed may be defective / not all seeds germinate any 1 to give a fair test	
			some seeds might not work	[1]
	(b)	(i)	add water to the cotton wool	[1]
		(ii)	all the oxygen has been absorbed / oxygen absorber (present)	
			without oxygen the cells cannot <u>respire</u> respiration is necessary to release energy energy is needed for growth / germination	[2]
	(iii)		temperature is too low / seeds are too cool / T is 4 °C	
			reactions are too slow at low temperatures reference to enzymes working slowly / inactive at low T	[2]

	Pa	ge 5		Syllabus	Paper
			GCE O LEVEL – October/November 2012	5129	22
11	(a)	(i)	I = V/R or 1.0/2 = 0.5		[2]
		(ii)	4.0		[1]
	(b)	(i)	reduced / decrease		[1]
		(ii)	reduced/ decrease		[1]
12	(a)	H ₂			[1]
	(b)	ste	c has gained oxygen am has lost oxygen cept correct explanations in terms of electrons or oxidation sta	ates	[2]
	(c)	(i)	oxygen / O ₂ water / H ₂ O accept steam / water vapour		[2]
		(ii)	galvanising do not accept: sacrificial protection / electrolysis		[1]
13	(a)	abs diffi cell allo	ey are soluble in water sorbed by root hair cell fusion I has large surface area (per volume) ow reference to active transport if given not accept: osmosis		[2]
	(b)	(i)	2200 (kg per hectare)		[1]
		(ii)	80 kg per hectare gives yield of 8200 kg per hectare 40 kg per hectare gives yield of 5900 kg per hectare 8200 – 5900 = 2300 kg per hectare (allow ecf for 1 mark if calculation is correct from incorrect re	eadings)	[2]
		(iii)	nitrogen (proteins) needed for growth nitrogen needed to make amino acids / proteins		[2]
		(iv)	9100 – 9200 (kg per hectare)		[1]
	(c)	pro pla (an	nversion of light energy into <u>chemical</u> energy oduction of carbohydrates / glucose ants are source of food / energy for animals need) oxygen (to breathe / respire) sintenance of O ₂ / CO ₂ balance in the atmosphere	2	[2]

	Page 6		Mark Scheme Syllabus			Paper	
			GCE O L	GCE O LEVEL – October/November 2012 5129		5129	22
14	(a)	C_nH_2	!n				[1]
	(b)	(i) addi	ion / reduction	on/ hydrogenati	ion / redox		[1]
		(ii) (carl	on to carbor	n) double bond	/ C=C		[1]
	(c)		Н	Н			
			c ——	- C) n			
			ı H	I H			
					es (below and after) (ind	ependent)	[2]
15	(a)	a blocka	ge of the (cor	onary) <u>arteries</u>			[1]
	(b)	high bloc lack of ex smoking	d pressure kercise life / life-style	lesterol diet / o	any 2		[2]
16	(a)	1.8					[1]
	(b)	9.16					[1]
17	anti anti	ckwise iclockwise iclockwise izontal / b	٦	all four correct			[2]
18	(a)	three sha					[2]
	(b)	covalent low non-meta	al non-met	al (1 mark for l	both)		[3]

	Page 7		Mark Scheme	Syllabus	Paper
			GCE O LEVEL – October/November 2012	5129	22
19	(a)	90 degre	es to mirror where ray is incident		[1]
	(b)	< incider		[1]	
	(c)	in approx		[1]	
20	(a)	carbon d	ioxide		[1]
	(b)	(b) acetylene and oxygen (both)			[1]
	(c)	nitrogen			[1]
	(d)	sulphur o	dioxide correct formulae		[1]